

SHORT  
COMMUNICATIONS

## Reaction of Tetracyanoethylene with 1,2-Cyclohexanedione and Bis(cyclohexanon-2-yl)methane

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The first study on reactions between tetracyanoethylene and carbonyl compounds was published in 1957 [1]. We extended the investigations in this field and established that the arising  $\beta,\beta,\gamma,\gamma$ -tetracyanoalkanes were very reactive [2, 3]. However in this process were involved only some  $\beta$ -dicarbonyl compounds whose addition to tetracyanoethylene followed by cyclization due to interactions  $\text{NH}\cdot\text{CN}$  and  $\text{OH}\cdot\text{CN}$  resulted in formation of pyrroles, furans, and pyrans [4].

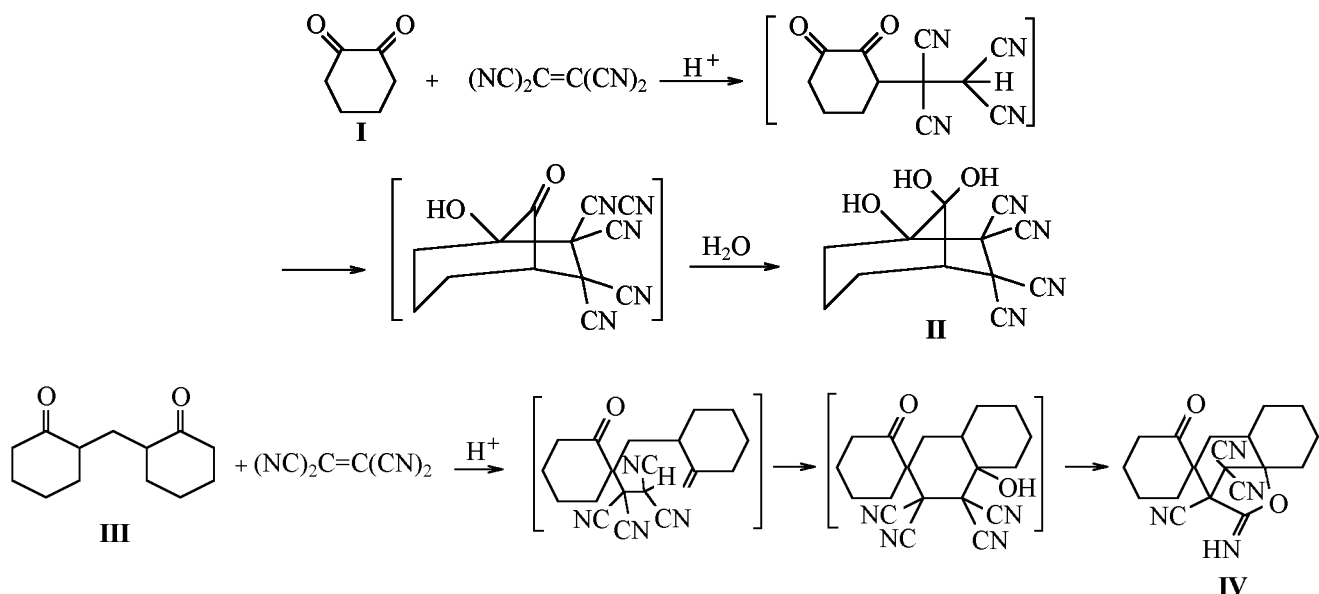
We found that the reaction between tetracyanoethylene and 1,2-cyclohexanedione (I) and bis(cyclohexanon-2-yl)methane (III) gave rise respectively to 6,6,7,7-tetracyanobicyclo[3.2.1]octane-1,8,8-triol (II) and 7-imino-2-spiro-1'-[2'-oxocyclohexane]-4,5-tetramethylene-6-oxabicyclo[3.2.1]octane-1,8,8-tricarbonitrile (IV). The main distinction of these reactions from those with  $\beta$ -dicarbonyl compounds consisted in

interaction of the acid fragment  $\text{CH}(\text{CN})_2$  with the carbonyl group resulting in formation of a cycloalkane ring.

In reaction of tetracyanoethylene with diketone I occurs unusual 1,3-cycloaddition to afford hydrate II, and with diketone III proceeds 1,4-cycloaddition to furnish hydrofuran IV.

The structure of compounds II and IV was determined by X-ray diffraction study.

**6,6,7,7-Tetracyanobicyclo[3.2.1]octane-1,8,8-triol (II).** To 1.28 g (0.01 mol) of tetracyanoethylene in 20 ml of dioxane was added 1.12 g (0.01 mol) of 1,2-cyclohexanedione and 5–6 drops of concn. HCl. In 24 h precipitate separated that was filtered off, washed with 2-propanol and ether. Yield 81%. mp 130–131°C. IR spectrum: 2260, 2270, 3100–3300  $\text{cm}^{-1}$ . X-Ray analysis data: space group  $P2_1/C$  at  $-90^\circ\text{C}$ ,  $a$  11.356(4),  $b$  8.596(3),



$c$  19.225(6) Å,  $\alpha$  90°,  $\beta$  90.73(3)°,  $\gamma$  90°,  $V$  1877(1) Å<sup>3</sup>,  $Z$  4,  $R$  0.045.

**7-Imino-2-spiro-1'-[2'-oxocyclohexane]-4,5-tetramethylene-6-oxabicyclo[3.2.1]octane-1,8,8-tricarbonitrile (IV).** To 1.28 g (0.01 mol) of tetracyanoethylene in 20 ml of dioxane was added 2.08 g (0.01 mol) of bis(cyclohexanon-2-yl)methane and 3–4 drops of conc. HCl. After 12 h was added 50 ml of water. separated a precipitate that was filtered off, washed with 2-propanol and ether. Yield 85%. mp 218–219°C. IR spectrum: 1700, 2260, 3280 cm<sup>-1</sup>. X-Ray analysis data: space group R2<sub>1</sub>/S at -90°C,  $a$  8.932(2),  $b$  10.008(5),  $c$  19.404(3) Å,  $\alpha$  90°,  $\beta$  93.56(10)°,  $\gamma$  90°,  $V$  1731.2(10) Å<sup>3</sup>,  $Z$  4,  $R$  0.0436.

## REFERENCES

1. Middleton, W.I., Heckert, R.E., Little, E.L., and Krespan, C.G., *J. Am. Chem. Soc.*, 1958, vol. 80, no. 11, pp. 2783–2788.
2. Nasakin, O.E., Sheverdov, V.P., Moiseeva, I.V., Lyshchikov, A.N., Ershov, O.V., and Nesterov, V.N., *Tetrahedron Lett.*, 1997, vol. 38, no. 25, pp. 4455–4456.
3. Nasakin, O.E., Sheverdov, V.P., Moiseeva, I.V., Ershov, O.V., Chernushkin, A.N., and Tafeenko, V.A., *Zh. Obshch. Khim.*, 1999, vol. 69, no. 2, pp. 302–311.
4. Sharanin, Yu.A., Goncharenko, M.P., and Litvinov, V.P., *Usp. Khim.*, 1998, vol. 67, no. 5, pp. 442–472.